PLOTNIKOV, V.G.; DANILOVA, W.I.; SHIGORIN, D.N.; TERPUGOVA, A.P.; ZUBKOVA, L.B.; FILIPPOVA, L.G.

Theoretical study of the spectral behavior of systems with a quasi-aromatic cycle. Zhur. fiz. khim. 39 no.9:2311-2312 S 165. (MIRA 18:10)

1. Institut neorganicheskoj khimii Sibirskogo otdeleniya AN SSSR.

	6016187			UR/0058/65/00U/011/1021/100
				.; Terpugova, A. F.
TITLE: Inve	stigation of lor recompounds	ng wave absorptio	n bands of ce	rtain polysubstituted //
SOURCE: Ref	. zh. Fizika, Al	s. 11D153		_
REF SOURC:	Tr. Komis. po sp	ektroskopii. AN	SSSR, t. 3,	vyp. 1, 1964, 345-349
TOPIC TAGS:	absorption band	, aromatic nitro	compound, or	ganic solvent, BENZENE
polysubstitu picric acid, ing the role gy levels of method. It compounds an currence of	tes of benzene (para nitrosophe of the NO2 grow the 2,4- and 2, is shown that the d the molecules a new absorption	para-nitrophenolenol, and 2,4-direction in the origin 6-dinitrophenolene hydrogen bond of the hydroxyl-	, 2,4-, 2,5-, itroaniline// of long-wave are calculat between the m containing so missing from	es of absorption of certai and 2,6-dinitrophenols, for the purpose of determi absorption bands. The ene ed by the free-electron olecules of the investigat lyents may lead to the oc- non-polar and oxygen-
SUB CODE:	07			
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Cord 1/1	JS			
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I, 32070-66 EWI(m)/EWP(j) RM ACC NR: AR6016174

SOURCE CODE: UN/0050/05/000/011/1015/1945

AUTHOR: Potapochkina, L. M.; Terpugova, A. F.; Zubkova, L. B.

TITLE: Investigation of singlet and triplet levels of anthraquinone and its derivatives

SOURCE: Ref. zh. Fizika, Abs. 11088

REF SOURCE: Tr. Komis. po spektroskopii. AN SSSR, t. 3, vyp. 1, 1964, 336-344

TOPIC TAGS: molecular orbital, molecular spectrum, nonmetallic organic derivative, luminescence quenching, hydrogen bonding, oxygen

ABSTRACT: Two methods (MO ICAO and MOSE) are used to calculate the energy spectrum and the wave functions of anthraquinone and some of its α - and β -derivatives. The α -derivatives of anthraquinone were calculated with and without allowance of the intramolecular H bond. Data are obtained on the influence of the structure and composition of the molecule, and also on the effect of the electron-donor properties of the substitute on the position of the singlet and triplet levels, making it possible to explain the experimental results of A. V. Karyakin, who investigated the fluorescence quenching of these compounds by oxygen [Translation of abstract]

SUB CODE: 20, 07

Card 1/1

JEALUGUEN, In. 1.

USSR/Cultivated Plants - Grains

M-4

Abs Jour : Ref Zhur - Riol., No 1, 1958, No 1481

Author

: M.P. Teremigova

Inst

: All-Union Agricultural Academy

Title

: The Morphological and Anatomical Characteristics of a Summer Duram Wheat Hybrid Produced by the VSKhI /All-Union Agricultural Institute, Department of Selection and Seed Culture.

Orig Pub : Zap. Vorozezhsk. s. kh. ir-ta, 1956, 26, No 2, 110-118

Abstract : There is a presentation of the results of an analysis of the hybrid, obtained by grafting "Melyanopus 69" wheat on the Persian wheat, which is distinguished by its vigor of development, fast ripening and hielding capacity. The peculiarities of the root development leaf system and stems have been studied. According to a number of characteristicsm the hybrid surpasses the original varieties of wheat, according to others, it takes an intermediate position or approaches the parantal forms.

Card : 1/1

FISHER, L.B.; TERPUGOVA, M.P.; KOTLYAREVSKIY, I.L.

Gatalytic dehydrogenation of 2,3-dimethylbutane. Izv. vost. fil.
AN SSSR no.9:53-56 '57.

1. Vostochno-Sibirskiy filial AN SSSR.
(Butane) (Dehydrogenation)

PISHER, L.B.; TERPUGOVA, M.P.; KOTHYAREVSKIY, I.L.

Dehydrogenation of butane di-derivatives; studying the first stage of 2.3-dimethylbutane dehydrogenation. Inv.Sib.otd. AN (MIRA 11:11)

SSSR no.9:32-38 '58. (MIRA 11:11)

1. Vostochno-Sibirskiy filial Akademii nauk SSSR. (Butane) (Dehydrogenation)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

"APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4 。 1. 1915年,1915年

SOV/153-2-4-26/32 5(1,3) AUTHORS:

Kotlyarevskiy, I. L., Fisher, L. B., Zanina, A. S., Terpugova,

M. P., Volkov, A. N., Shvartsberg, M. S.

TITLE: Synthesis of Several Monomers on Alumochromium Catalysts

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Khimiya i khimicheskaya

tekhnologiya, 1959, Vol 2, Nr 4, pp 608 - 613 (USSR)

ABSTRACT: A report on this paper was given at the All-Union Conference on "Ways of Synthesis of Initial Products for the Production of High Polymers" which took place in Yaroslavl' from September 29 to October 2, 1958. The results of the catalytic synthesis of 2,3-dimethyl-butadiene-1,3 as well as of monomers of the type of p-xylene from acetylene derivatives are given. The substance mentioned at first can be used in the production of special rubber types (Ref 1). It could not be obtained by the production methods so far used. The diagram (see Diagram) being worked at by the authors consists of two stages. The first one (Ref 2) yielded not more than 15% of the end product. In spite of numerous patents (Ref 3), a thorough description of reaction

conditions is still missing. Therefore, the authors determined Card 1/4 the optimum conditions of isobutane alkylation with ethylene

Synthesis of Several Monomers on Alumochronium Catalysts 507/153-2-4-26/32

in the presence of aluminum chloride. Under these conditions $(48 - 51^{\circ}, 7 - 9 \text{ atm}, 1.5 \text{ kg isobutane}, 145 \text{ g ethylene}, 40 \text{ g}$ C_2H_5C1 , 5 g AlCl₃, 2 hours) the yield of 2,3-dimethyl-butane rises to 47%. The reaction is very sensitive to temperature (only 1/3 of the alkylate yield at 35°). The catalyst can be used 5 to 6 times without reducing the quantity of alkylate or of 2,3-dimethyl-butane. Ethylene has to be added during the whole process: its partial pressure must not exceed 0.5 - 0.7 atm, or otherwise the 2,3-dimethyl-butane content in the alkylate decreases rapidly. The addition of 1-3% C2H5Cl accelerates the process. Isobutane alkylation with ethylene at a high temperature proceeds at a pressure of 100 - 200 atm according to a radical mechanism, and is accelerated by radical donors. It was carried out by the authors in the presence of $\mathrm{C_{2}H_{5}Cl}$ (for the first time) in a special, continuously working plant. The alkylate yield (computed for ethylene) reached 170-180% under optimum conditions (450°, 200 atm, reaction time 25 minutes, weight ratio isobutane : ethylene = 10 : 1, initiator quantity

Card 2/4

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

2.5%). 2,2-Dimethyl-butane (32-35% of the alkylate) was the

Synthesis of Several Monomers on Alumochromium Catalysts SOV/153-2-4-26/32

main product; 2,3-dimethyl-butane amounted to 8-10% of the alkylate. The preparation of 2,3-dimethyl-butadiene-1,3 from 2,3-dimethyl-butane proceeds in two stages: a) Dehydration with the formation of two olefines: 1) tetramethyl-ethylene, and 2) 2,3-dimethyl-butene-1. b) Dehydration of the olefines to the end product. This reaction has not been considered in publications (except for the reference in the patent (Ref 5)). The authors investigated the first stage thoroughly on an alumochromium-potassium catalyst. The specifications for its production were provided by M. N. Marushkin (deceased) (IOKh AN SSSR = Institute of Organic Chemistry AS USSR). This catalyst showed the highest activity at 550°. The catalyzate reached a 87% yield with a content of unsaturated hydrocarbons of 60-67%. The investigation of the second stage has not yet been completed. An aromatization diagram of divinyl-acetylene-hydrocarbons is given. Several mono-, bi-, and tricyclic hydrocarbons with a prescribed structure were produced on the basis of the diagram. A diagram of the reactions of the synthesis of p-xylene from acetylene and acetone is suggested. Since there is no demand for acetone in the USA because of increasing phenol production,

Card 3/4

Synthesis of Several Monomers on Alumochronium Catalysts SOV/153-2-4-26/32

and its price according to the Sergeyev method is already 1/3 - 1/4 of the present one, the synthesis method mentioned above might become useful for industry. In conclusion, a simple way of synthesis for polyphenyl systems is suggested. Moreover, the Favorskiy reaction is mentioned in the paper. There are 11 references, 5 of which are Soviet.

ASSOCIATION: Vostochno-Sibirskiy filink SO AN SSER (East Siberian Branch of the Siberian Department of the Academy of Sciences, July)

Card 4/4

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

PISHER, L.B.; TERPUGOVA, M.P.; KOTLYAREVSKIY, I.L.

Dehydrogenation of disubstituted butanes. Dehydrogenation of 2,3-dimethylbutane and 2,3-dimethylbutanes in the presence of 2,3-dimethylbutane and 2,3-dimethylbutanes. [MIRA 14:3) diluents. 1xv.Sib.ofd.AN SSSR no.2:57-60 '61. (MIRA 14:3)

1. Vostochno-Sibirskiy filial Sibirskogo otdeleniya AN SSSR, Irkutsk. (Butane) (Butene) (Dehydrogenation)

KALABINA, A.V.; TYUKAVKINA, N.A.; TERPUGOVA, M.F.

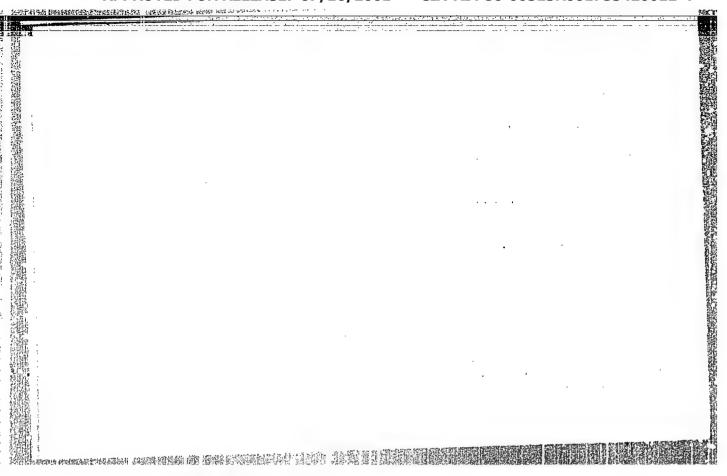
Synthesis and some properties of d, p-dichloroethyl ethers of the aromatic series. Izv.vys.ucheb.zav. khim.i khim.tekh. 4 no.4:632-635 '61. (MIRA 15:1)

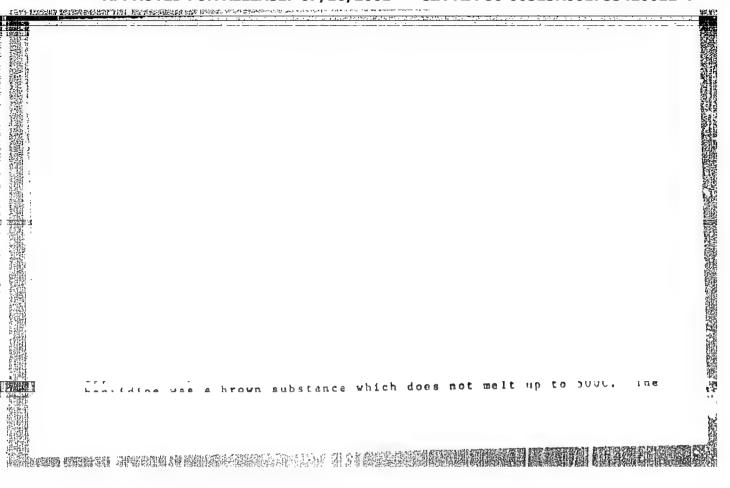
1. Irkutskiy gosudarstvenny universitet imeni Zhdanova, kafedra vysokomolekulyarnykh soyedineniy i organicheskogo sinteza. (Ethers)

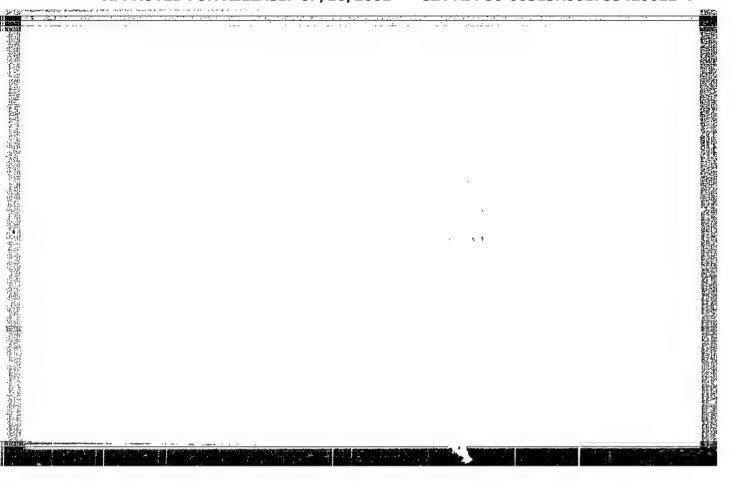
KOTLYAHEVSKIY, I.L.; TERPUGOVA, M.P.; ANDRIYEVSKAYA, E.K.

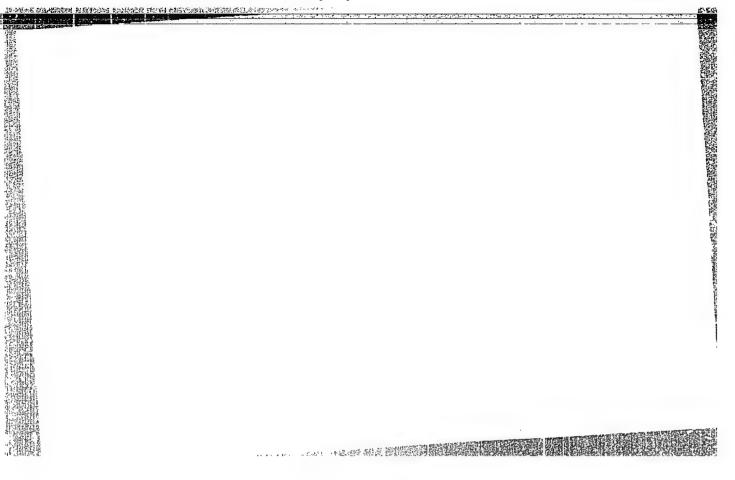
Highly unsaturated polymers. Report No.10: Polymers having azo groups in the chain. Tzv. AN SSSR. Ser. khim. no.10:1854-1860 0 '64. (MIRA 17:12)

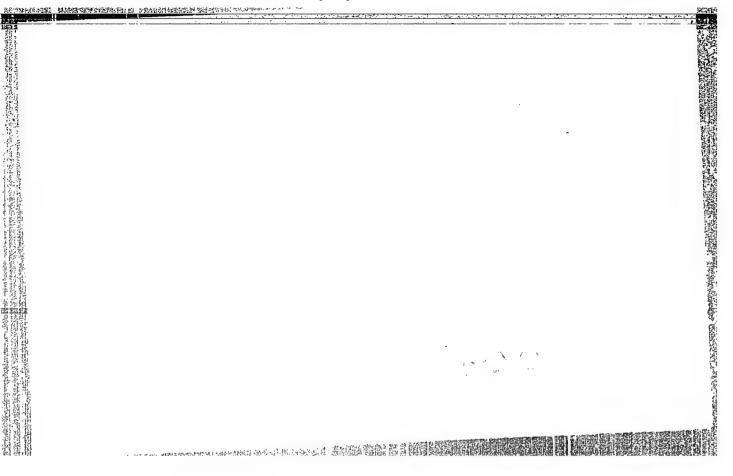
1. Institut khimicheskoy kinetiki i goreniya Sibirskogo otdeleniya AN SSSR.

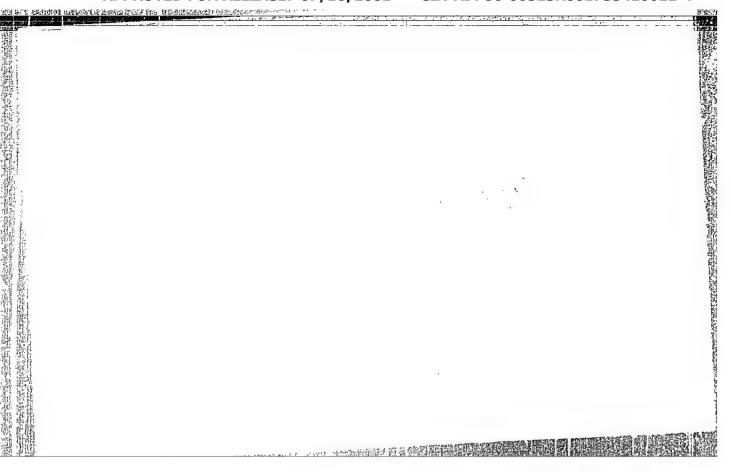


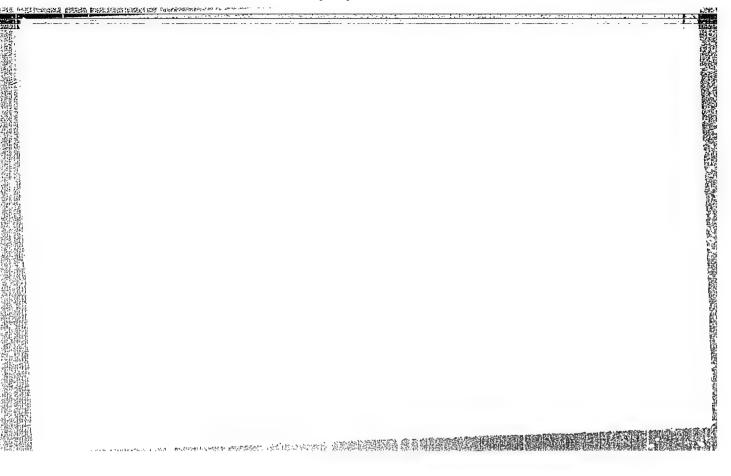


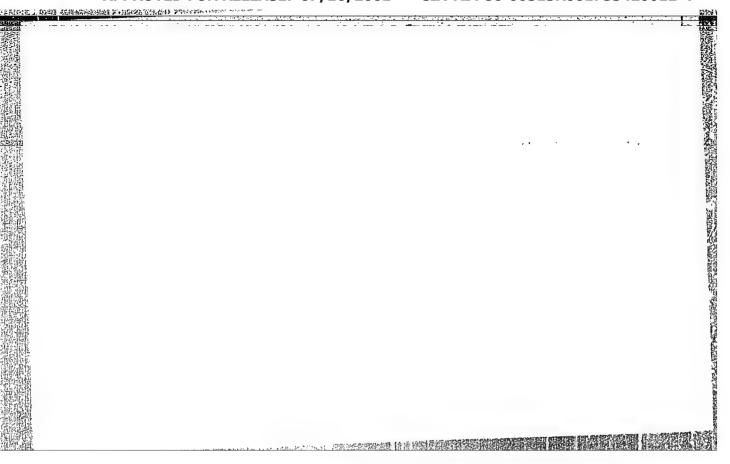


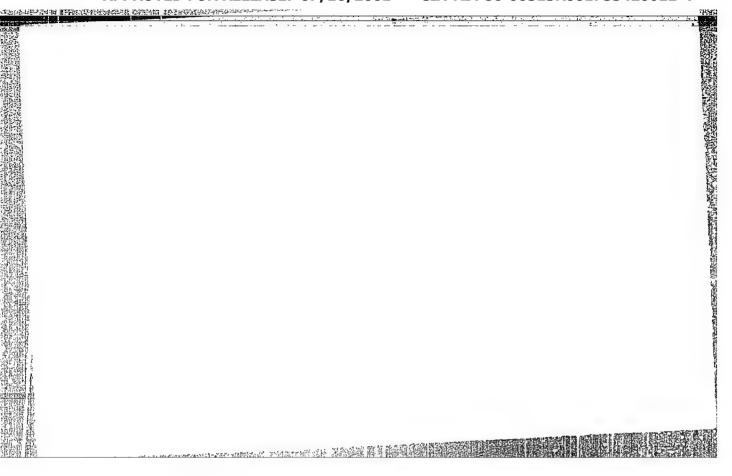


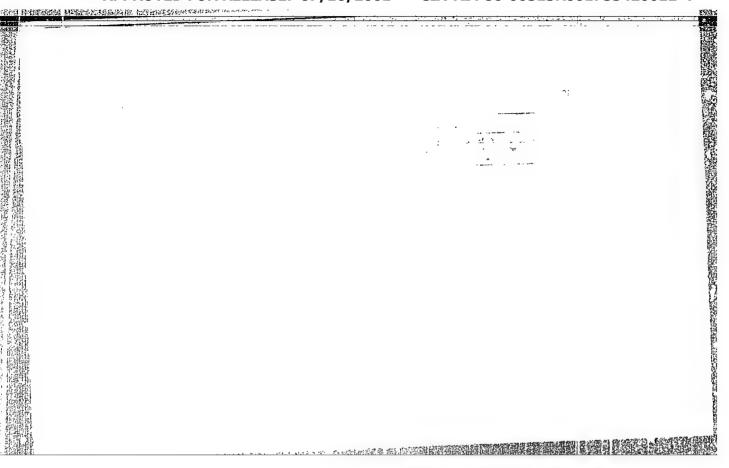












SCHOOL DESCRIPTION OF THE PROPERTY OF THE PROP

KOTLYAPEVSKIY, I.L.; TERFUGOVA, M.P.; MITY SCHOVA, A.A.

Synthesis of diphenylpicrylhydrazyl. fzv. SO AN SSSR no.2 Ser. khim. nauk no.1:151-152 165. (MISA 18:8)

1. Institut khimicheskoy kinetiki i goreniya Sibirskego otdeleniya AN SSSR, Novosibirsk.

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

TOTAL TERMINE AND THE TOTAL PROJECT TOTAL PROJECT TOTAL TOTAL TOTAL TOTAL TOTAL TOTAL PROJECT TOTAL TOTAL PROJECT TOTAL SOURCE CODE: UR/0062/66/000/004/0713/0720 45 EVI (m) /ENP(1)/I 23866-66 AUTHOR: Terpugova, H. P.; Kotlyarevskiy, I. L.; Andriyevskaya, E. K.B. ACC NR. AP6014409 ORG: Institute of Chemical Kinetics and Combustion, Siberian Department of the Academy of Sciences SSSR (Institut khimicheskoy kinetiki i goreniya Sibirakogo otdeleniya Akademii nauk SSSR) TITLE: Highly unsaturated polymers. Communication 15. Synthesis and some physical properties of polyaropolyarenes? AN SSSR. Izvestiya. Seriya khimicheskaya, no. 4, 1966, SOURCE : TOPIC TAGS: organic semiconductor, semiconducting polymer, polyazo-713-720 polyarene, oxidative polycondensation, electric property ABSTRACT: New homo- and co-polymeric polyazopolyarenes have been prepared and their physical and electrical properties investigated. This work was part of a systematic study of the effect of the structure of highly unsaturated polymers on their properties. The polymers had the general formula, H2N-Ar-H(-H-Ar'-H-H-Ar-H)n-H-Ar'-HH2, 2 Card 1/2

L 23866-66 3 ACC NR. AP6014409 where Ar and Ar' may be identical or different. The homo- and copolymers (listed in the source) were prepared by oxidative polycondensation of aromatic dismines in pyridine solution in the presence of CuCl. The diamines used were o-tolidine, bis (p-aminophenyl)methane, and 4,4'-diaminostilbene. In addition, o-phenylenediamine was used, which should not form straight-chain polymers, and (p-aminophenyl) acetylene, which should form polymers containing both are and butadiyne groups in the backbone. Butadiyue groups should form cross-links on heating, thereby improving electrical conductivity. These dismines and (p-aminophenyl) acetylene were homopolymerized and copolymerized with each other and with p-phenylenediamine, benzidine, and chrysoidine. The polymer structures were confirmed by elemental analysis and IR spectroscopy, and showed an EPR signal. Elemental analysis and IR spectra revealed partial oxidation to form -N+0 bonds. homo- and co-polymers vere fusible and more soluble in chloroform, tetrahydrofuran, acetone, and dioxane than the infusible (straightchain] polymers. The room temperature conductivity of all the polymers was low, 10-13 to 10-14 mho/cm, but rose rapidly with temperature, reaching 10^{-8} to 10^{-7} mho/cm for some of the polymers at 200-250C. Some of the polymers exhibited a very high activation energy for conduction, 2-3.5 ev. Orig. art. has: 3 tables and 1 figure. SUBH DATE: 18Nov63/ ORIG REF: 002/ OTH REF: 004 SUB CODE: 07. 11/ ATD PRESS: 42 46 Card 2/2440

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

L 5118-66 EFT(1)/T LJP(c)

ACC NR: AP5025086

SOURCE CODE: UR/0368/65/003/003/0209/0216

。 國面特理等的最後的數据的自己的數据等的數据,在1945人的形式。 1960年

AUTHORS: Freebrazhenskiy, N. G.; Ravodina, O. V.; Terpugova, N. S.

ORG: none

7/44/>>
TITLE: Formation of spectral line shape with asymmetrical self-reversal

SOURCE: Zhurnal prikladnoy spektroskopii, v. 3, no. 3, 1965, 209-216

TOPIC TAGS: spectrometry, line self reverse, spectral line intensity, spectrum line, spectrum analysis

ABSTRACT: The present paper is a continuation of work reported previously by N. S. Gorbachava and N. G. Preobrazhenskiy (Opt. i spektr., 15, 453, 1963). The factors determining the spectral line shape with asymmetrical self-reversal were investigated. It was found that the line shape could be described in terms of four parameters ϕ_1 , ϕ_2 , ϕ_3 , and ϕ_4 (see Fig. 1). The relationships between these parameters and the optical density and inhomogeneity of the radiating layer, the Voigt parameter, and the shifts and half-widths of emission and absorption lines

Card 1/2

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L 5418-66 ACC NR: AP5025086

Fig. 1 Parameters in terms of which the line shape I (x) is described. $\phi_1 = I_{M_2}/I_{M_2}, \phi_2 = I_m/I_{M_2}, \phi_3$ line width measured at half-line width of the smaller maximum, \$\distance between maxima

were determined. The results are given in tabular form. Orig. art. has: 9 tables, 1 graph, and 13 equations.

SUB CODE: OP/

SUBM DATE: 12Jan65/

ORIG REF: 005/

OTH REF: 005

BVK. Card 2/2

ACC NR: AP7003146

SOURCE CODE: UR/0368/66/005/006/0706/0711

AUTHOR: Preobrazhenskiy, N. G.; Kolobova, G. A.; Terpugova, N. S.

ORG: none

TITLE: Theory of quantitative spectrum analysis with a laser excitation source

SOURCE: Zhurnal prikladnoy spektroskopii, v. 5, no. 6, 1966, 706-711

TOPIC TAGS: laser application, spectrum analysis, quantitative analysis, optic density, laser spectroscopy

ABSTRACT: The extensive inhomogeneity and considerable optical density characteristic of the luminous layer produced by using a laser to heat a specimen make conventional methods for recording the integral line intensity unsuitable. The spectral region separating the self-reversed maxima is preferable as a measure of the concentration of the element in question. The paper contains a theoretical study of the dependence of the above spectral region on the optical thickness of the emitting layer under various conditions of spectrum excitation. Orig. art. has: 19 formulas and 2 figures. [Authors' abstract]

SUB CODE: 20/SUBM DATE: 26Jul65/ORIG REF: 007/OTH REF: 006/

表。 第一次,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是一个人的,我们就是

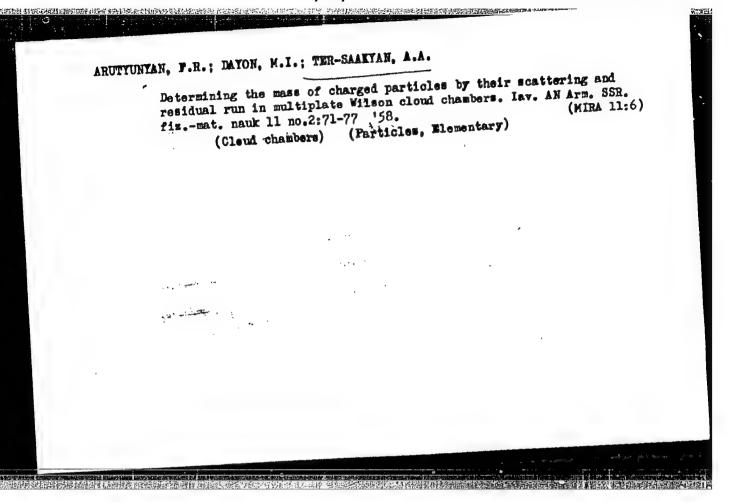
Cord 1/1 UDC: 543. 42

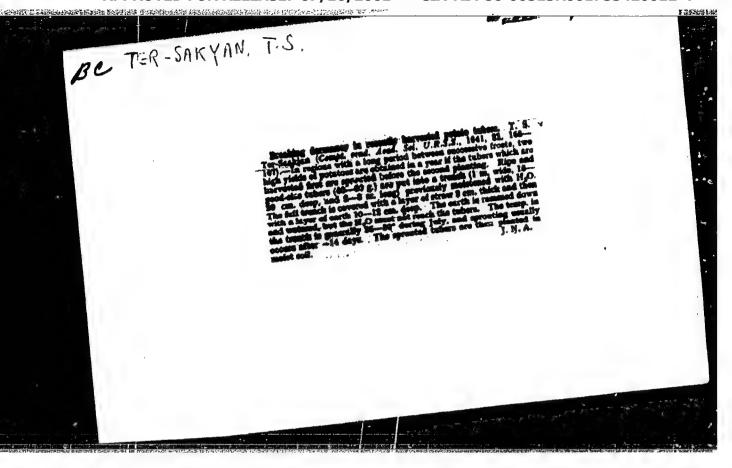
TERRY, C.A., TINSLEY, A. B.; ZARKOVIC, Stjepan (translator)

Arc welding of low-alloy steel plates with the application of powder on the front and back sides of the weld. Zavarivanje 3 no. 7/8:143-149 S-0 '60.

KASUMZADE, N.G.; TER-SAAKOV, B.U.; MAMEDOV, N.A.; ARAKELOV, A.S.; SPEKTOR, Sh.Sh.; NEGREYEV, V.F., red.; ZEYNALOVA, T.Z., red. izd-va; AKHMEDOV, S., tekhn. red.

[Protection of apparatus and equipment of petroleum refineries from corrosion]Zashchita apparatury i oborudovaniia neftepererabatyvaiushchikh zavodov ot korrozii. [By] N.G. Kasumzade i dr. Baku, Azerneshr. 1962. 282 p. (MIRA15:9) (Petroleum refineries—Equipment and supplies) (Corrosion and anticorrosives)



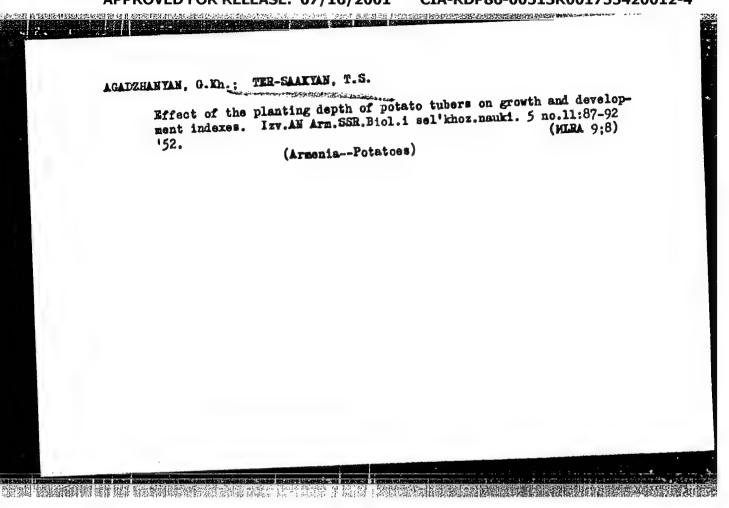


Mor., Inst. Farm Studies, Dept. Agric. Sci., America Acad. Sci., 12.3-47.

Mor., Inst. Farm Studies, Dept. Agric. Sci., America Acad. Sci., 12.3-47.

Cand. Agricultura Sci.

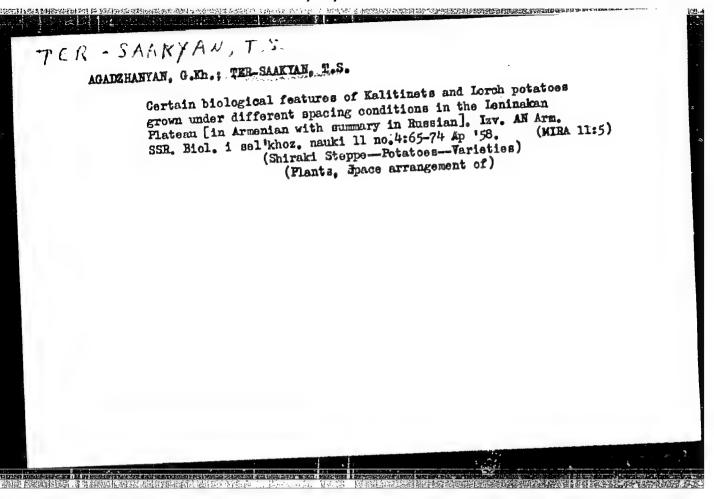
"Two Potato Herveste from the Same Tubers," Lok. AM, 41, No. 1, 1943.

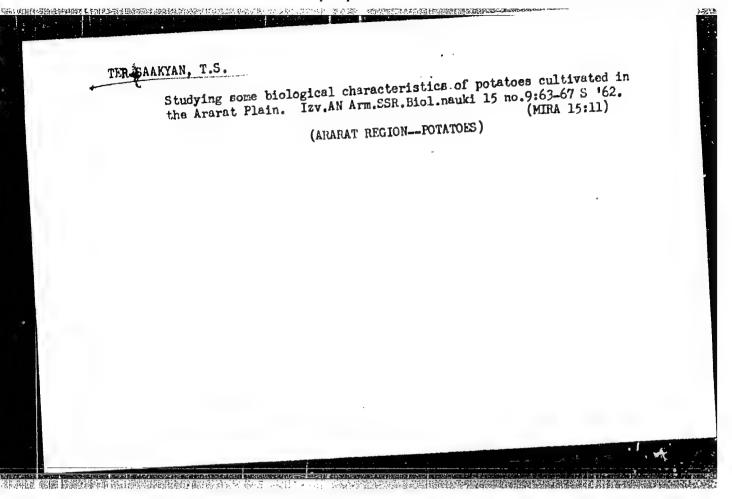


AGADZHANYAN, G.Kh.; TER-SAAKYAN, T.S.

Effect of planting time on the yield and quality of the potato crop
[in Armenian with summary in Bussian]. Isv.AN Arm. SSR. Biol. 1
[in Armenian with 3.125-36 Kr '54.
eel'khoz.nauki 7 no.3125-36 Kr '54.
(Shiraki Steppe--Potatoes)

Burn UATEGORY ABS. JOUR. : RZD101., No. 1959, No. 10313 AUTHOR : Ter-Saakyan, T. 6.737 Tribula Siera Servorm of the Potato and New Method of Controlling h ORIG. PUB. : Ayastani koltnicsakan, Kolkhoznik Armenii, 1958, Mo 4, 49-30 ABSTRACT No abstract. JARD: 1/1





TERSAR, B.

Putting a new article into production of organized electric industry. p. 388.

ELEKTROTEHNISKI VESTNIK. ELECTROTECHNICAL REVIEW. Ljubljana, Yugoslavia. Vol 26, no. 11/12, 1958.

Monthly List of East European Accessions (EEAI) LC, Vol. 8, no. 6, June 1959. Uncl.

WELLY TO BE RECEIVED THE RESIDENCE AND THE PARTY OF

ALIYEV, Vagab Safarofich; AL'TMAN, Natal'ya Borisovna; TER-SARKISOV, Ben'yamin Georgiyevich; NAGIYEV, M.F., akademik, red.;
BAGDATLISHVILI, D., red. izd-va; POGOSOV, V., tekhn. red.

[Research in the field of catalytic and thermalcontact refining of the heavy fraction of crude] Issledovaniia v oblasti kataliticheskoi i termokontaktnoi pererabotki tiazhelogo neftianogo syr'ia. Baku, Izd-vo Akad. nauk Azerbaidzhanskoi (MIRA 15:4) SSR, 1961. 282 p.

(Petroleum-Refining)

ALIYEV, V.S.; KASIMOVA, A.P.; TER-SARKISOV, B.G.

Method of determining the activity of K-5 finely divided
catalysts. Azerb. neft. khoz. 40 no.9:33-35 S '61. (MIRA 15:1)
(Catalysts)

ALIYEV, V.S.; YEFIMOVA, S.A.; KASIMOVA, A.P.; TER-SARKISOV, 3.G.

Evaluation of the activity of catalysts used in industrial processes with a circulating powdered catalyst. Kin.i kat. 3 no.4: 545-549 Jl-Ag '62. (MIRA 15:8)

1. Institut neftekhimicheskikh protsessov AN Azerbaydzhanskoy SSR. (Catalysts)

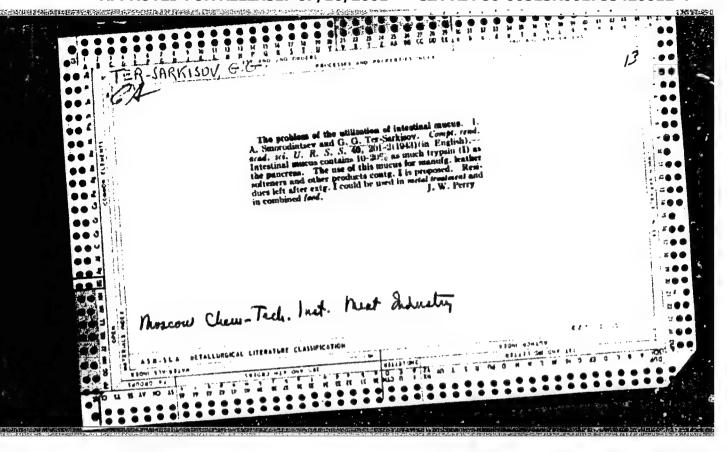
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ALIYEV, V.S.; ALIYEV, Z.E.; KASIMOVA, A.P.; KAPIANOVA, V.D.; MURAVCHIK, M.Ye.; TER_SARKISOV, B.G.

Preliminary preparation of the dehydrating K-5 catalyst before its introduction into the reactor. Azerb.neft.khoz. 41 no.8: (MIRA 16:1) 35-39 Ag 162. (Catalysts)

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"



TER-SARKISOV, R., khudozhnik-konstruktor

Industrial interior decoration. Tekh. est. 2 no.7:16-17 J1 '65. (MIRA 18:8)

1. Spetsial noye khudozhestvenno-konstruktorskoye byuro Leningradskogo sovetu narodnogo khozyaystva.

SOURCE CODE: UR/0251/66/042/003/0547/0550 EWT(1)/EEC(k)-2/TIJP(c) 46120-66 AUTHOR: Gogava, L. A.; Nakashidze, G. A.; Delerzon, N. M.; Dzhaparidze, Ye. G.; AP6024547 Kakhabrishvili, I. V.; Ter-Sarkisova, A. G.

ORG: Academy of Sciences, Georgian SSR, Institute of Cybernetics (Akademiya nauk Gruzinskoy SSR, Institut kibernetiki)

TITLE: Photoelectric characteristics of a two-terminal p-n-p-n type transistor switch

SOURCE: AN GruzSSR. Soobshcheniya, v. 42, no. 3, 1966, 547-550

TOPIC TAGS: electronic switch, germanium transistor, photosensitivity, volt ampere characteristic, projunction, photosectric propurty

ABSTRACT: The article deals with the method of fabrication and photoelectric characteristics of germanium-base p-n-p-n type transistor switches. The starting material was a p-type wafer with a resistivity of 5 ohms cm and dimensions of 1.3x1.3x0.08 mm. Two p-n junctions were obtained by diffusing antimony into both surfaces of the original wafer and the third, by alloying indium into one of the diffused layers. Ohmic contact on the opposite side was accomplished by doping with tin (Fig. 1). In the presence of a fixed bias lower than the switching

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L 4612C-66 ACC NR: AP6024547

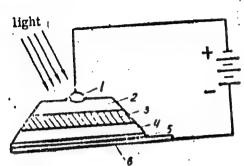
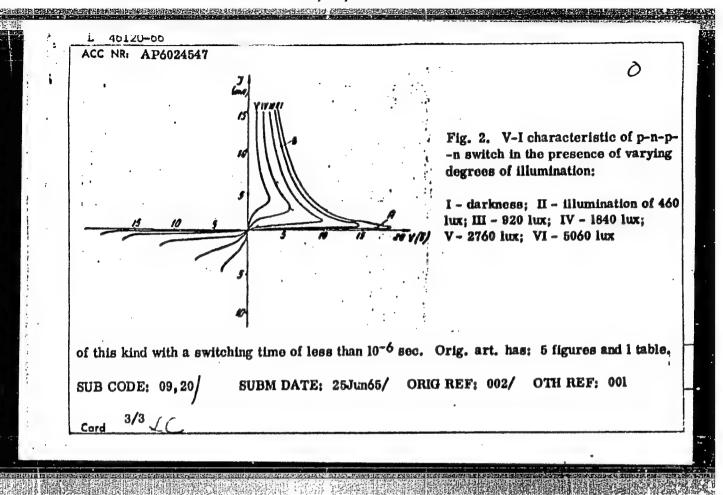


Fig. 1. Structure of two-terminal p-n-p-n type switch:

1 - rectifying nickel contact;
2,4 - diffused
n-layers;
3 - original p-type germanium;
5 - ohmic contact (tin);
6 - nickel holder

voltage the device is in the "off" state (point A on V-I characteristic in Fig. 2) and displays a high resistance of the order of several megohms. On illumination the switch changes from "off" state to "on" state (point B in Fig. 2) considering that the fixed bias voltage is then sufficient for breakdown of the center p-n junction. In this position the resistance of the desufficient for breakdown of the center p-n junction. In this position the resistance of the desurge is of the order of several ohms. An investigation of V-I characteristics in the presence of darkness and various degrees of illumination conclusively proved that switching voltage decreases with increasing illumination. The minimum illumination required to switch the decreases with order of 100-150 lux. Further improvements in the design and fabrication of vice is of the order of 100-150 lux. Further improvements in the design and stable devices transistor switches should make it possible to develop more photosensitive and stable devices

Card 2/3



\$/079/62/032/009/011/011 1048/1242

是一个文字,对于2世中在1000年2000年2000年200日,1000年200日的1000年201日,1000年201日,1000年201日,1000年201日

AUTHORS:

Zhivukhin, S.M., Dudikova, E.D., and Ter-Sarkisyan, E.M.

TITIE:

Synthesis and investigation of organostannoxanes. II

PERIODICAL: Zhurnal obshchey khimli, v.32, no.9, 1962, 3059-3061

TEXT: The first paper on this subject appeared in this publication, v. 31, 1961, 3106. This paper reports the results of an attempt to prepare polyorganestrannoxanes by polycondensation of dibutyldiace-toxystannane and dibutyldibutoxystannane according to the equation

The experiments were carried out in an inert gas stream with constant stirring at 150 to 2000C. The starting materials were syn-

Card 1/2

s/070/62/032/000/011/011 1048/1242

Synthesis and investigation ...

thosized in the laboratory. Both the rate of condensation and the final yield of butyl acetate increased, while the molecular weight of the polymer decreased with increasing reaction temperature. Thus, at 150°C the reaction was completed within 2 hrs, the yield of butyl acetate was 61.4%, and the molecular weight of the polymer was 1290 when the reaction was carried out under atmospheric pressure and 2400 in vacuo. At 180°C the reaction was completed within 1.5 hrs, the yield of butyl acetate was about 75%, and the molecular weight of the polymer was 840 under atmospheric pressure and 1610 in vacuo. The product formed at temperatures above 200°C contained a non-melting, insoluble phase whose structure needs further clarification. The polymer formed at 180°C was a brittle substance, easily soluble in benzene and its homologs, with a softening temperature of 130-140°C. The polymer formed at 150°C was a glass-like, brittle and transparent substance with good adhesion to metals and glass; its m.p. was 96-100°, and it did not lose its melting characteristics on repeated melting-solidification cycles. There are 2 figures.

SUBMITTED: September 13, 1961

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

TER-SARKISYAN, G. S.	methylscridine mpl, contrary to expectations, lowers capacity of W group to react in condensations with aromatic aldehydes, p-nitrosodimethylsniline, and aromatic aldehydes, p-nitrosodimethylsniline, and alazo compds. In these reactions quaternary salts of I exhibit much greater mobility of H atoms in We of I exhibit much greater mobility of H atoms in We aroup than free base. Azo dyestum prepd by reacting I with p-nitrodiszobenzene is identical with p-nitrophenylhydrazone of 9-methyl-3, 4-benzoscridyl p-nitrophenylhydrazone of 9-methyl-3, 4-benzoscridyl aldehyde.	USSR/Chemistry - Aeridine Deriva- tives (Contd)	pp 601-609 synthesized series of new derivs of 9-methyl- synthesize	Actions Commu Commu Benzoacr	USSR/Chemistry - Aeridine Deriva- Sep/Oct
PA 195T21	ons with e, and y salts y salts ms in We y react with hzoscridyl	Sep/oct 51	hyl- duction	or stor	72

TER-SARKISYAN, G. S.

USSR/Chemistry - Acridine Derivatives

Hov/Doc 51

"Condensation of Benzoderivatives of 9-Methylacridine. II. Transformations of 9-Methyl-1,2-Benzoacridine (I) and 9-Methyl-3, h, 5,6-Dibenzoacridine (II), A. Ye. Poray-Koshits (Deceased), G. S. Ter-Sarkisyan, Inst Urg Chem, Acad Sci USSk

"Iz Ak Nuak SSSR, Otdel Khim Nauk" No 6, pp 771-776

Introduction of a condensed benzene nucleus into 9-methylacridine lowers the reactivity of the methyl group in condensations with aromatic aldehydes, nitroso compds, and diazo compds. New de-

PA 197T13

TIR-SARKISYAN, G. S. TFR-SARKISYAN, G. S. -- "Condensation Reactions of Genzelots (Genzeline or Tenzene Analogs) of L Nethyl-Pyridine." Sub 26 Dec 32, Into of ir ship Chemistry, Acad Sci USSR. (Dis.ertation for the Degree of Tandidate in

Chemical Sciences).

Vechernaya Moskva January-December 1952 SO:

CIA-RDP86-00513R001755420012-4" APPROVED FOR RELEASE: 07/16/2001

USSR / chemistry digards chemistry 1 1/1 Pub. 40 - 12/27 Card Mikhaylov, B. M., and Ter-Sarkisyan, G. S. Authors | landensation reactions of nonzepontratives of tempthylactifing. Title Concensation with , - ... to sectimethy last the Periodical : Izv. AN ESSR. Otd. khim. nauk 4, 656 - 662, July - August 1954 . The contensation of customarritime, metryl-1, -benzacritime Abstract water and the second of the se LIBES IN the molecule. Lue sliser of arrigatores fals? 'n am ram or condensation reaction, is explained. It was found that 9 methylacridine condenses easily with p-nitrockdimethylaniline in darkness, at room Institution : Acad. of Sc. USSR, The W. D. Zelinskiy Institute of Organic Chemistry Submitted : June 23, 1953

TER-SAKKISVAR ES

USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 901

Author: Mikhaylov, B. M., and Ter-Sarkisyan, G. S.

Institution: Academy of Sciences USSR

Title: Relative Reactivity of the Methyl Group in the Benzene Homologs of

4-Methylpyridine

Periodical: Izv. AN SSSR, Section on Chemical Sciences, 1954, No 5, 846-853

Abstract: The reactivity of the CH3-group in picoline (I), lepidine (II), 5,6-

benzolepidine (III), 7,8-benzolepidine (IV), 9-methylacridine (V), 9-methyl-1,2-benzacridine (VI), 9-methyl-3,4-benzacridine (VII), 9methyl-1,2,7,8-dibenzacridine (VIII), and 9-methyl-3,4,5,6-dibenzacridine (IX) has been studied in the condensation with m-NO2C6H4CHO (X). It has been established that the reactivity of the investigated compounds in the reaction is expressed by the series I > II > II > IV and V > VI > VII > VIII > IX and, as a result, that the mobility of the hydrogen of the CH3-group in the benzene homologs of I depends

Card 1/3

USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 901

Abstract: both on the number and on the position of the benzene nuclei condensed with the pyridine molecule. A mixture of 0.19 mole of A. B-dinaphtylamine, 0.19 mole (CH3CO)2O (XI), and 26 gms anhydrous ZnCl2 is heated for 5 hours at 185-1900. The CHqCOOH is distilled off and the residue heated 30 minutes at 250-2600, followed by repeated treatment with 10% H_SOh and neutralization with 25% NHhOH; VIII is obtained in yields of 46.8% (crude), mp 180-1830 (successive crystallization from benzene, ethylacetate, and alcohol). Chromatographic purification of crude VIII yields an isomer with mp 215-2160. A mixture of 2.7 moles of II, and 5 ml of XI is heated in a sealed tube for 1.5 hours at 1.5 hours at 150-1530; the contents of the tube are dissolved in 20 ml C6H6. The solution is treated with 30 ml 6 N HCl; 4-(m-nitrostyryl)-pyridine is obtained by the neutralization of the HCl-solution (yield, 48.2%); the unreacted II is recovered as the semicarbazone after extraction with benzene (yield 47.85%, based on II charged). Similar procedures were used in the condensation of the above-named benzene homologs of I with X and of II with o-NO₂C₆H₄CHO. A mixture of 2.2 moles of the hydrochloride of III, 2.2 moles of X, and 1.5 mole of XI is refluxed 3 hours; 10 ml of water are added after cooling and the solution is

Card 2/3

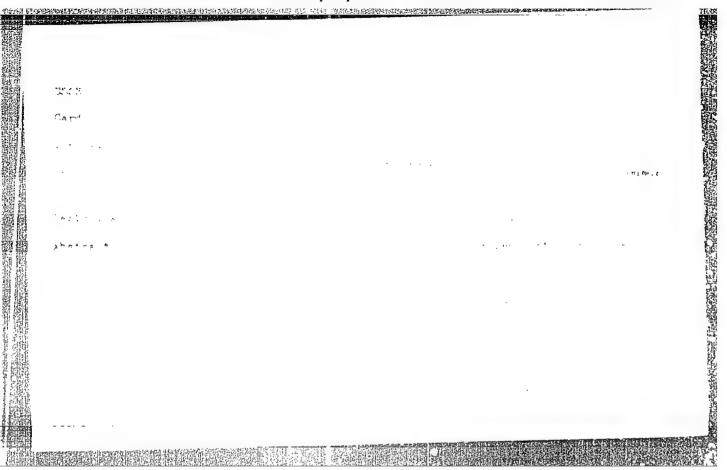
USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

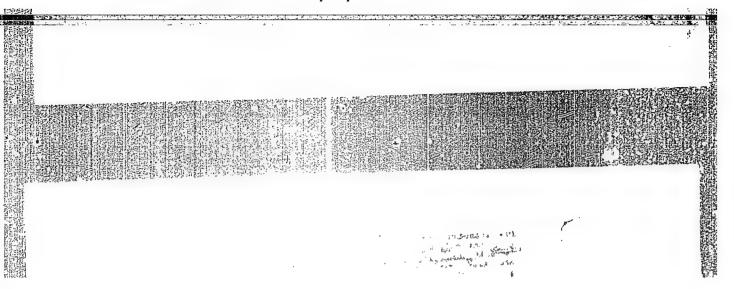
Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 901

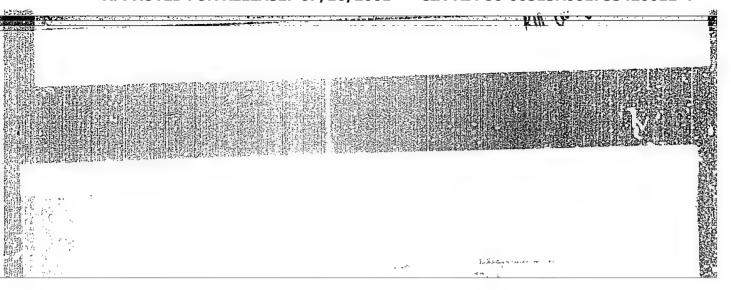
Abstract: made alkaline with 25% NH4OH. The precipitate is washed twice with 3-ml portions of alcohol; the yield of m-nitro-benzylidene-5,6benzolepidine is 30%, mp 168-1690 (from alcohol). Similar methods were used in the synthesis of n-dimethyl-aminobenzylidene-5,6benzolepidene in yields of 26.9%, mp 170.5-171.50 (from alcohol); m-nitrobenzylidene-7,8-benzolepidine in yields of 21.8%, mp 155-157° (successive crystallization from methyl and ethyl alcohol); and 9-(n-dimethylaminostyry1)-2-methylacridine in yields of 94% (crude), mp 225-2260 (successive crystallization from CH3OH and a mixture of benzene and petroleum ether). Condensation of V and X under UV-irradiation yields α-(m-nitrophenyl)-β-(9-acrydil)-ethanol (XII): a mixture of 7 mmoles of V, 7 mmoles of X, and 14 ml of C6H6 is irradiated with UV light for 100 hours in a N2-atmosphere; the yield of XII is 54%, mp 146.5-147.5° (successive crystallization from benzene and dioxane). Similar methods were used in the condensation of X with II, III, and VI.

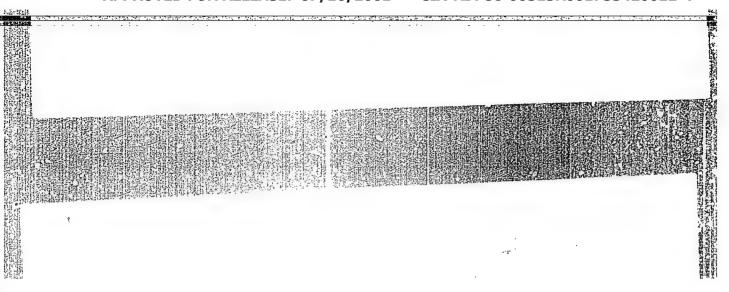
Card 3/3

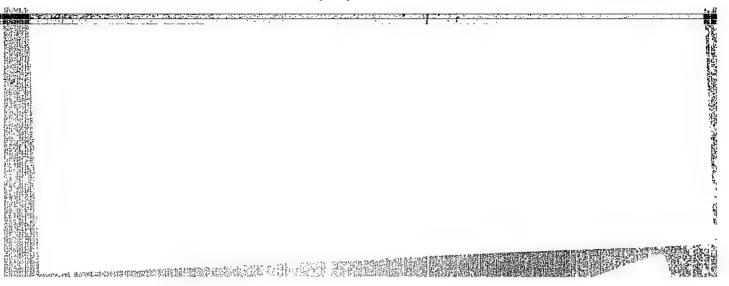
CIA-RDP86-00513R001755420012-4" APPROVED FOR RELEASE: 07/16/2001

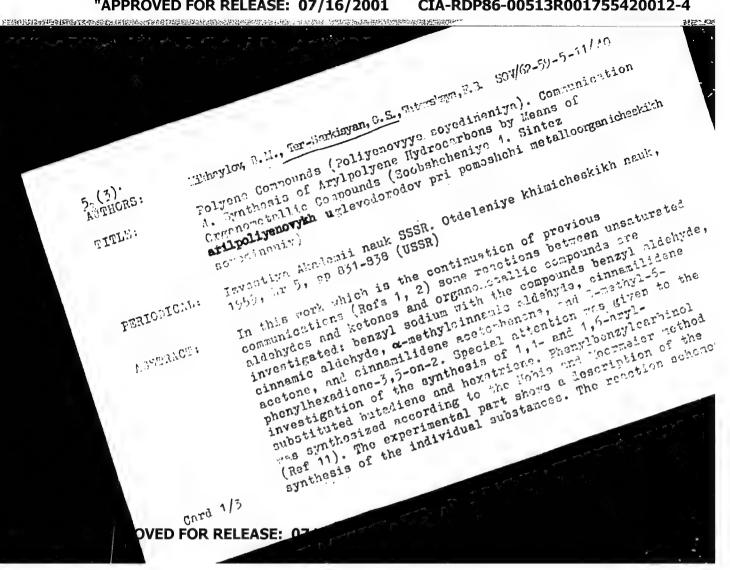












Polyene Compounds. Communication 4. Synthesis of SCY/62-59-5-11/62 Arylpolyene Hydrocarbons by Means of Organometallic Compounds

for the preparation of the individual compounds are given. In the synthesis benzyl sodium was attached to cianable alteral in 1,2-position and methylcianamic aldehyde as well as a unsaturated ketone which has a methyl group in addition to carbonyl group. The attachment of benzyl sodium to cianable acetophone took place in the position 1,4 whereas the large ketone reacted with expicolyl lithium in 1,2-position. Parcyl lithium reacted with benzaldehyde and cianamilidane acetomy with the formation of carbinols; in connection with the hydrogenation of the latter, dimers of the corresponding hydrocarbons were formed. Moreover, new forms of polyenus, e.g. the compound (XVIII) were obtained. The following compounds had been synthesized earlier: (VI) by Bower (126 %, (IX) by Koler (Ref 5), (dimerization of butadienc know, literature (Refs 6, 7), Whitby (Ref 8)), (XIX) by refere, 10.

Card 2/2

Polyene Compounds. Communication 4. Synthesis of S0V/62-59-5-11/40 Arylpolyene Hydrocarbons by Teans of Organometallic Compounds

ASSOCIATION: Institut organicheskoy khimii im. N. D. Zelinskogo Akademii

nauk SSSR (Institute of Organic Chemistry imeni N. D.

Zelinskiy of the Academy of Sciences, USSR)

SUBMITTED: July 18, 1957

Card 3/5

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

SOV/51-6-3-12/28

AUTHORS: Nikitina, A.N., Galanin, M.D., Ter-Sarkisyan, G.S., and Mikhaylov, B.M.

TITLE: The Absorption and Luminescence Spectra of Solutions of Substituted Polyenes (Spektry pogloshcheniya i lyuminestsentsiya rastvorov nekotorykh zameshchennykh poliyenov)

PERIODICAL: Optika i Spektroskopiya, 1959, Vol 6, Nr 3, pp 354-365, (USSR)

ABSTRACT: The authors investigated the electronic absorption spectra of eighteen substituted butadienes and hexatrienes dissolved in heptane as well as luminescence of solutions of these All the substances substances in heptane and benzene. studied were purified chromatographically using aluminium The absorption spectra of solutions were measured The luminescence spectra oxide. using a spectrophotometer SF-4. in the visible region were measured by means of a spectrometer consisting of a monochromator UM-2 and a photo-The results obtained are shown in Table multiplier FEU-19. This table includes calculated values of the oscillator Card 1/2 strengths of long-wavelength electronic transitions and the

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

SOV/51-6-3-12/28

The Absorption and Luminescence Spectra of Solutions of Substituted Polyenes

quantum yields of luminescence. The absorption spectra of solutions of the substituted butadienes and hexatrienes are shown in Figs.1-8. It was found that the absorption intensities and band positions depend on the degree of intensities and band positions depend on the degree of intensities and band positions depend on the degree of intensities and band positions depend on the degree of intensities and band positions and in other of some was found also that the quantum yield of luminescence of some was found also that the quantum yield of substances is higher in benzene solutions and in others it is substances is higher in benzene solutions of the excited-higher in heptane solutions. Measurements of the excited-higher in heptane solutions. Measurements of the excited-state lifetime of the quantum yield of state lifetime showed that decrease of the quantum yield of state lifetime showed the second type, while changes due primarily to quenching of the second type, while changes of the excited-state lifetime of 1,6-diphenyl-hexatriene-1,3,5 of the excited-state lifetime of 1,6-diphenyl-h

SUBMITTED: January 16, 1958

Card 2/2

ACC NR: AP6009804 Dibutyl ester of allylboronic acid gives high yields of 2-allyl-1,4-hydroquinone(I) Dibutyl ester of allylboronic acid gives high yields of 2-allyl-1,4-hydroquinone(I) which otherwise is hard to obtain. Butyl ester of diallylborinic acid and triallyl which otherwise is hard to obtain. Butyl ester of diallylborinic acid and triallyl which otherwise is hard to obtain. Butyl ester of diallylborinic acid and triallyl boron yield a mixture of (I) and 2,5-diallylphenol (II). The following data were boron yield a mixture of (I) and 2,5-diallylphenol (II). The following data were determined for (II): bp = 114-115C (4.5 mm); n _D ²⁰ = 1.5400; and d ₄ ²⁰ = 0.9966.								
determin	ned for (II)	: bp = 114-	-115C (4.5	OTH REF: O			(EH	
B02 002		••		,	·	٠.,		
		•	<i>:</i>		***			
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	•	,			•			

5(3) AUTHORS: Mikhaylov, B. M., Ter-Sarkisyan, G. S. SOV/79-29-5-50/75

TITLE:

Polyene Compounds (Poliyenovyye soyedineniya).
VII. Condensation of Vinyl-Ethyl-Ether With the Acetals of Aromatic Aldehydes and Ketones (Kondensatsiya viniletilovogo efira s astetalyami aromaticheskikh al'degidov i ketonov)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 5, pp 1642-1648

(USSR)

ABSTRACT:

As a first example in the series of aliphatic-aromatic ketones the authors investigated the reaction between diethyl ketal of acetophenone and the vinyl-ethyl-ether. This reaction is much more complicated than the one between acetals of aromatic aldehydes and α,β -unsaturated ethers. The following was isolated from the mixture: 3-phenyl-1,1,3-triethoxy butane, 5-phenyl-1,1,5-triethoxy hexene-2, 7-phenyl-1,1,7-triethoxy octadiene-2,4,9-phenyl-1,1,9-triethoxy-decatriene-2,4,6 and othyl acetal of acetal aldehyde. The structure of 3-phenyl-1,1,3-triethoxy butane was proven by saponification with 1,1,3-triethoxy butane was proven by saponification with diluted hydrochloric acid to 3-phenyl-3-ethoxy-butanal-1 and, under heavier conditions, to β -methyl cinnamaldehyde.

Card 1/2

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

Polyene Compounds. VII. Condensation of Vinyl-Ethyl- SOV/79-29-5-50/75 Ether With the Acetals of Aromatic Aldehydes and Ketones

In order to determine the position of the ethoxy group in 5-phenyl-1,1,5-triethoxy hexene-2, the diethyl acetal of β-methyl cinnamaldehyde was entered into reaction with vinyl ethyl ether and 5-phenyl-1,1,3-triethoxy hexene-4 was obtained therefrom. Both hexene derivatives (hexene-2, hexene-4) were hydrolyzed and the aldehyde obtained was identified by its 2,4-dinitrophenyl hydrazone. Moreover, hexene-2-derivative was obtained by direct synthesis from acetophenone acetal and 1-ethoxy butadiene-1,3. Both higher condensation products were not investigated more closely. The experimental describes the reactions carried out and gives the physical and analytical data of the initial substances as well as of the compounds obtained. There are 8 references, 1 of which is Soviet.

ASSOCIATION:

Institut organicheskoy khimii Akademii nauk SSSR (Institute of Organic Chemistry of the Academy of Sciences of the USSR)

SUBMITTED:

April 15, 1958

Card 2/2

SOV/79-29-8-22/81

5(3) AUTHORS:

Mikhaylov, A. M., Ter-Sarkisyan, G. S.

TITLE:

Polyene Compounds. IX. Tondensation of Acetals of Furyl-, Cinnamic - and Furyl-acrylic Aldehyde With Unsaturated Ethers

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 8, pp 2560-2565

(USSR)

ABSTRACT:

In addition to previous papers (Refs 1,2), the present paper deals with the condensation of the diethyl acetals of the abovementioned aldehydes with the vinyl ethyl ether and 1-ethoxybutadiene-1,3. The reaction of these diethyl acetals with α,β butadiene-1,3. The reaction of these diethyl acetals with α,β butadiene-1,3. The reaction of these diethyl acetals with α,β butadiene-1,3. The reaction of these diethyl acetals with α,β butadiene-1,3. The reaction of the presence of the acetic acid solution of zinc chloride, and yields the corresponding ethoxy derivatives of the aromatic and furane series which, by boiling with acetic acid (Ref 3), are easily transformed into the unsaturated aldehydes. The reactions of the diethyl acetals of the above-mentioned aldehydes with vinyl ethyl ether gave, in fair yields, the condensation products of acetal, mostly with one molecule of vinyl ethyl ether already at equimolar ratios of the reagents. Thus, the diethyl acetal of the cinnamic aldehyde reacts with this ether under formation of

Card 1/3

907/79-29-8-22/81

Polyene Compounds. IX. Condengation of Acetals of Furyl-Cinnamic- and Furylacrylic Aldehyde With Unsaturated Ethers

compound (I) which is transformed, by boiling with glacial acetic acid, to give (II) in good yield. This method is of preparative importance, and better than the previous condensation according to reference 4 (Scheme 1). Compound (III) can be obtained as easily, which is transformed by boiling with acetic acid into the acrolein (IV) which had formerly been synthesized from furfurole and acetic anhydride (Ref 5) (Scheme 2). The reaction of compound (V) with vinyl ethyl ether gives (VI), and further, with acetic acid, pentadienal (VII) (Scheme 3). On reaction of the double quantity of diethyl acetal of the cinnamic aldehyde with 1-ethoxy-butadiene-1,3, compound (VIII) resulted which further gives compound (IX) by boiling with acetic acid (Scheme 4). Further compounds of this kind (X)-(XIII) were synthesized for the first time (Schemes 5,6,7). There are 7 references, 2 of which are Soviet.

ASSOCIATION:

Institut organicheskoy khimii Akademii nauk SSSR (Institute of Organic Chemistry of the Academy of Sciences, USSR)

Card 2/3

5/062/60/000/007/013/017/XX BO04/B064 and Ter-Sarkisyan, G. S. Polyene Compounds. Communication 11. Asymmetrical Mikhaylov, B. M., 1,6-Diaryl-substituted Hexatrienes-1,3,5 AUTHORS: PERIODICAL: Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh TITLE: In a previous paper, the authors have described a simple method of synthesizing 5-phenyl pentadien-2, 4-al-; from the diacetal method of synthesizing 5-phenyl pentadien-2, 4-al-; from the diacetal of cinnamic acid and vinyl ethyl ether. Thus, this aldehyde became easily accessible to various syntheses, especially to the synthesis of commentation o easily accessions to various syntheses, especially to the syntheses. In of asymmetrical 1,6-disubstituted hexatriene-1,3,5 derivatives. In the present page the cuthors access to the contract of the cuthors access to the contract of the cuthors access to the cuthor the present paper, the authors report on the syntheses carried out the present paper, the authors report on the syntheses carried out (A) with 5-phenyl pentadien-2,4-al-1, 1,6-diphenyl hexatriene-1,3,5 (A) with p-pnenyl pentagren-2,4-almi, 1,0-dipnenyl nexarriene-1,
a) by means of the Grignard reagent obtained from magnesium and a) by means of the Grignard reagent obtained from magnesium and benzyl chloride. The reaction is carried out in absolute ether benzyl chloride. The reaction is carried out in toluene (yield: 31.5%); (yield: 26.5%); b) with benzyl sodium in toluene (yield: 31.5%); Card 1/3

Polyene Compounds. Communication 1:Asymmetrical 1,6-Diaryl-substituted
Hexatrienes-1,3,5

S/062/60/000/007/013/017/XX B004/B064

c) with phenyl acetic acid in acetanhydride in the presence of lithargo and boiling in nitrogen atmosphere, yield: 34.5%. 2) :-pheny:-6- $(\alpha-naphthyl)-hexatriene-1,3,5$ (B) with $\alpha-naphthyl$ acetic acid in acetaldehyde and in the presence of litharge; y'eld: 20.4%. Contrary to A, this compound luminesces strongly both in crystallized and dissolved state. 3) 1-phenyl-6 (9-phenanthryl)-hexatriene-1,3,5 (C) by means of the Grignard reagent from Mg and 9-shloro methyl phenanthrene in ether. This compound luminesces. 4) 1-phenyl-5hydroxy-6-(p-hiphenyl)-hexadiene-1,3 (D) by means of the Grignari reagent from Mg and 4-phenyl benzyl chloride in absolute ether. 5) 1-phenyl-6-(p-biphenyl)-hexatriene-1,3,5 (E) by boiling of D in glacial acetic acid. 6) 1-phenyl-6-(2-pyridyl-hexatriene-:,3,5 (F) with α -picolyl-lithium in absolute ether. 7) 1-phenyl-5-hydroxy-6-(2-quinoly1) hexadiene-2.4 (G) with quinaldy1-lithium in ether. 8) 1-phenyl-6-(2-quinolyl)-hexatriene-1,3,5 (H) by boiling of G in acetic acid. The introduction of heterocyclic substituents (E and G) leads to a weaker luminescence. In a later paper the authors will report on the optical properties of these compounds.

Card 2/3

Polyene Compounds, Communication 1; Asymmetrical 1,6-Diaryl-substituted

S/062/60/000/007/013/017/XX B004/B064

other determination of the property of the control of the control

Hexatrienes-1,3,5

There are 5 references: 3 Soviet, 1 US, and 1 Swiss.

ASSOCIATION:

Institut organicheskoy khimii im. N. D. Zelinskogo

Akademii nauk SSSR

(Institute of Organic Chemistry imeni N. D. Zelinskiy

of the Academy of Sciences USSR)

SUBMITTED:

December 27, :958

Card 3/3

CIA-RDP86-00513R001755420012-4" APPROVED FOR RELEASE: 07/16/2001

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2209, 1150,1300

S/062/60/000/010/029/031/XX B004/B060

... HORS:

Mikhaylov, B. M. and Ter-Sarkisyan, G. S.

7:25E:

Synthesis of Mercaptals

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk,

1960. No. 10, pp. 1886-1887

TEXT: The recently much investigated reaction of aldehyde acetals with vinyl ethers meets with difficulties in some cases, because of the complicated character of synthesis of corresponding acetals. In the study under consideration the authors intended to simplify their task by using easily obtainable mercaptals, RCH(SC₂H₅)₂, instead of acetals. The following restance is described: A mixture of other mercaptan with ZnCl and Na-SO.

action is described: A mixture of ethyl mercaptan with ZnCl₂ and Na₂SO₄ is cooled down to -2°C, and aldehyde is added dropwise. The temperature must not exceed 0°C. After being allowed to stand in a refrigerator for +0 h, the mixture is poured into ice water, the separating oil is extacted with ether, washed with 10% lye, and subsequently with water, dried over Na₂SO₄, and distilled in vacuum. Under these mild conditions, diethyl Card 1/2

Synthesis of Mercaptals

86718 \$/062/60/000/010/029/031/XX B004/B060

mercaptals of cinnamaldehyde, 5-phenyl pentadien—2,4-al, citral, β-cyclocitral, benzaldehyde, and furfurole were obtained in good yields (55.1-77.5%). A report is to follow concerning the condensation of these mercaptals with α,β-unsaturated ethers. L. S. Povarov is mentioned. There are 1 table and 10 references: 5 Soviet, 2 US, 1 Japanese, and 2 Swiss.

ASSOCIATION:

Institut organicheskoy khimii im. N. D. Zelinskogo Akademii

nauk SSSR

(Institute of Organic Chemistry imeni N. D. Zelinskiy of the

Academy of Sciences USSR)

SUBMITTED:

March 21, 1960

Card 2/2

0071y

-3832

2209, 1234, 1274

S/062/60/000/010/030/031/XX B004/B060

AN HORS:

Mikhaylov, B. M. and Ter Sarkisvan, G. S.

TIE:

Condensation of Mercaptals With Vinyl Ethyl Ether

FERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk,

1960, No. 10, pp. 1888-1891

TEXT: In a previous paper (Ref. 2), the authors synthesized diethyl mercaptals of benzaldehyde, cinnamaldehyde, and β -cyclocitral. The article under consideration deals with the condensation of these compounds with vinyl ethyl ether. The reaction took place with mercaptal in the presence of BFz stherate or ZnCl2 solution in ethyl acetate as a catalyst, addition by

drops of vinyl ether without allowing temperature to rise, mixing with ether, and distillation of the ether extract in vacuum. The following scheme is

 $RCH(SC_2H_5)_2 + CH_2 = CHOC_2H_5 \rightarrow RCH(SC_2H_5)CH_2CH SC_2H_5$; R = (I); CH_3 CH_3 (III).

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Condensation of Mercaptals With Vinyl Ethyl Ether S/062/60/000/010/030/031/XX B004/B060

The presence of the mercaptal group was proved by a sublimate (Ref. 5). Boiling with glacial acetic acid effects the conversion of synthesized compounds into unsaturated aldehydes. Cinnamaldehyde was obtained from 1-ethoxy-1,3di-(ethyl mercapto)-3-phenyl propane (I), while 5-phenyl pentadien-2,4-al was obtained from 1-ethoxy-1,3-di-(ethyl mercapto)-5-phenyl pentene-4 (II). Saponification of (I) in the presence of 2,4-dinitro phenyl hydrazone (2,4 DNPH) gave 2,4-DNPH of 3-phenyl-3-ethyl mercapto propanal, which, on boiling with alcoholic HCl, was converted into 2,4-DNPH of cinnamaldehyde. The ethyl mercapto group, not the ethoxy group, is separated on further reaction of (I) with vinyl ethyl ether to form 1,3-diethoxy-1,5-di-(ethyl mercapto)-5-phenyl pentane. The structure of this compound was proved a) by reaction with sublimate (proof of the mercapto group); b) reaction with 2,4-DNPH: formation of 2,4-DNPH of 5-phenyl-5-ethyl mercapto penten-2-al, which was converted into 2,4-DNPH of 5-phenyl pentadien-2,4-al. There are 1 table and 7 references: 5 Soviet, 1 US, and 1 French.

ASSOCIATION:

Institut organicheskoy khimii im. N. D. Zelinskogo Akademii

nauk SSSR (Institute of Organic Chemistry imeni N. D. Zelinskiy

of the Academy of Sciences USSR)

SUBMITTED:

March 23, 1960

Card 2/2

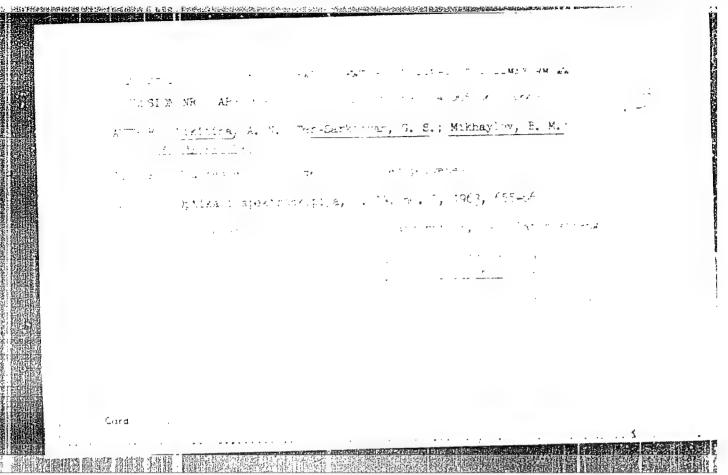
MIKHAYLOV, B.M.; TER-SARKISYAN, G.S.

Polyene compounds. Fart 12: Condensation of 5-pheny1-2,4-pentadienal with arylacetic acids. Zhur.ob.khim. 30 no.8:2521-2524 Ag '60.

(MIRA 13:8)

1. Institut organicheskoy khimii Akademii nauk SSSR.

(Acetic acid) (Pentadienal)



L 9849-63 ACCESSION IC: AP3000505

long wavelength electronic transitions were calculated. The experimental late are table, and if the experimental late are table, in section of an analyses of increasence years of a terminal are characteristic table memoers of the nexactricite series. The department of a terminal area of the experimental area of the nexactricity series of the experimental area of the nexactricity of the experimental area of the experimental area of the experimental area.

ASSOCIATION: none

SUBMITTED: BAUG62 DATE ACQ: 12Jun63 ENCL: 00

SUB CODE: CH, PH NR REF SOV: CG8 OTHER: 003

nh/ja Card 2/2

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

MIKHAYLOV, B.M.; TER-SARKISYAN, G.S.; BYSTROVA, A.A.

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Polyenic compounds. Report No.16: Thicketals of unsaturated ketones and their transformations. Izv.AN SSSR. Ser.khim. no.1:46-50 Ja '64. (MIRA 17:4)

1. Institut organicheskoy khimii im. N.D. Zelinskogo AN SSSR.

Hinting, L.M., Francaskinski, O.S.; Mikhashov, B.M.; Minchenkov,

Fluorescence of solution of substituted polyenes. Acta
physica Pol 26 no.3/4:483-487 S-O '64.

1. Organic Chemistry Institute, Moscow.

MIKHAYLOV, B.M.; TER-SARKISYAN, G.S.; BYSTROVA, A.A.

Folyene compounds. Report No.17: Condensation of vinyl ethyl ether with thic ketals. Izv. AN SSSR. Ser. khim, no.3:443-446 (MIRA 18:5)

1. Institut organicheskoy khimii im. N.D. Zelinskogo AN SSSR.

TER-SARKISYAN, G.S.; MIKHAYLOV, B.M.

Condensation of unsaturated thio ethers of the terpone series with orthoformic ester. Izv. AN SSSR. Ser. khim. no.3:561-563 *65. (MIRA 13:5)

1. Institut organicheskoy khimii im. N.D.Zelinskogo AN SSSR.

MIKHAYLOV, B.M.; TER-SARKISYAN, G.S.

Polyene compounds. Report No. 18: Conversions of ethylmercapto-substituted monothicacetals of ∞ - and β -dihydroionylidenacetaldehydes. Izv. AN SSSR. Ser. khim. no.7:1197-1204 165. (MIRA 18:7)

1. Institut organicheskoy khimii im. N.D. Zelinskogo AN SSSR.

TER-SARKISYAN, G.S.; MIKHAYLOV, B.M.

Reaction of 1-(2,6,6-trimethylcyclohexen-2-yl)-3-(ethylmercapto)1,3-butadiene

with thicacetals. Zhur. org. khim. 1 no.7:1239-1241 Jl 65. (MERA 18:11)

1. Institut organicheskoy khimii imeni N.D.Zelinskogo AN SESR.

APPROVED FOR RELEASE: 07/16/2001 CIA-RDP86-00513R001755420012-4"

MISHUKOVA, Ye.A.; TER-SARKISYAN, L.G.

Electrophoresis of sarcoplasmic proteins in the rabbit myccardium under normal condition; and in experimental diphtheria
intoxication. Vop. med. khim. 9 no.20204-206 Mr. ap '63.

(MIR' 17:8)

1. Kafedra biskhimii zhivotnykh Meskovskogo gosudarstvernogo
universiteta imeni lomonosova.

COTREL, Y.; TERSEN, G.; MOREL, G.

Corrective fixation in scoliesis. Acta chir. orthop. traum. cech. 29 no.5:393-395 0 162. (SCOLIOSIS) (BONE TRANSPLANTATION)

TERSENOV ,S .A.

Rigenvalues and eigenfunctions of vibrations in a cylindrical envelope. Soob.AE Grus. SSE 15 no.9:575-581 154.

(MLRA 8:9)

1. Akademiya nguk Grusinskoy SSR, Tbilisskiy matematicheskiy institut im. A.M.Rasmadze. Predstavleno deystvitel'nym chlenom Akademii I.N.Vekua (Elastic plates and shells) (Eigenfunctions)

124-57-2-2242

Translation from: Referativnyy zhurnal, Mekhanika, 1957, Nr 2, p 111 (USSR)

AUTHOR: Tersenov, S. A.

TITLE: The Asymptotic Behavior of Eigenvalues and Eigenfunctions of

the Vibrations of Cylindrical Shells (Asimptoticheskoye povedenive sobstvennykh znacheniy i sobstvennykh funktsiy kolebaniy

tsilindricheskikh obolochek)

PERIODICAL: Soobshch. AN GruzSSR, 1955, Vol 16, Nr 1, pp 11-18

ABSTRACT: Utilizing Carleman's method, asymptotic formulas are derived

for the eigenvalues and eigenfunctions of the system of equations for the stationary vibrations of a circular cylindrical shell

₁λ_{U=0},

with the following boundary conditions:

u=0, v=0, w=0, $\frac{\partial w}{\partial y}=0$ at L.

Here U is a vector function having u, v, and w as its components; L is the boundary of the region D of the median surface of the shell; Y is the normal to L; P is a matrix operation with

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124-57-2-2242

The Asymptotic Behavior of Eigenvalues and Eigenfunctions (cont.)

components ℓ_{ij} ; and λ is the frequency parameter. These equations take into account the moments of the inertia forces which usually are neglected. As is shown by the asymptotic formulas, the moments of the inertia forces affect the frequencies of the higher vibratory modes significantly. The author

$$\lim_{n\to\infty} \ \frac{n}{\lambda_n} = \left(\frac{2b+c}{4\pi b(b+c)} + \frac{1}{8\pi (b+c)}\right) I,$$

where I is the area of the region D, and b and c are constant quantities which depend on the Poisson ratio. The last term takes the moments of the inertia forces into account. According to the author, the difference $\lambda_n^* - \lambda_n$ (where λ_n is the value that obtains when the moment of the inertia forces is disregarded) for extremely high frequencies exceeds 10 percent of λ_n and ultimately attains values between 10 and 16 percent of λ_n

1. Cylindrical shells--Mathematical analysis

O. D. Oniashvili

Card 2/2

TERSENOV, SA

20-4-10/60

AUTHOR:

Tersenov, S.A.

TITLE:

An Elliptical Type of Equation which Degenerates on the Boundary of the Domain (Ob odnom uravnenii ellipticheskogo tipa vyrozhdayushchemsya na granitse oblasti)

PERIODICAL:

Doklady Akademii Nauk SSSR, 1957, Vol. 115, Nr 4, pp. 670-673 (USSR)

ABSTRACT:

The author investigates the equation L(u) # Ju + u + au + au + bu + cu = 0. In this connection the a(x,y), b(x,y) and c(x,y) in any finite portion of the semiplane y > 0 are analytical functions of the independent variables x and y, and lytical functions of the independent variables x and y, and c (x,y) ≤ 0 may apply. In the semiplane y > 0 the initially given equation is of an elliptical type and it degenerates at y = 0. The author here gives an in a certain sense/constructive characteristic of the solutions of the above-mentioned equation near the boundary of the degeneration of the time dependent investigates the boundary conditions in a formulation suggested by A.V. Bitsadze. The author investigates the function $\omega(x,y) = \binom{4}{2} \exp \left(\binom{4}{3}(x,r) r^{-1} dr\right) dt + C_0$. The constant

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An Elliptical Type of Equation which Degenerates on the Boundary of the Do-

C is selected so that $\omega > 0$ holds in the closed domain \overline{D} . The function ω (x,y) is an analytical function for the finite quantities x and y > 0. Two theorems are given and proved here: Theorem 1: For any continuous function f(x,y) assumed on f(x,y) assumed on f(x,y) of the initially given equation which satisfies the condition f(x,y) = f(x,y)

 $(x,y) \rightarrow 0$ $Q \in \Gamma + AB$. In this connection AB signifies a section of the axis 0x and Γ signifies a smooth open arc. Theorem 2: When u(x,y) is a twofold continuous differentiable solution of the initially described equation in D, when the boundary values on the closed set \overline{G}_0 are continuous and when $\lim_{x \to \infty} (u/\omega) = 0$

applies on G, such a solution is limited in \overline{D} . This solution is then in a unique manner defined by the assumption of continuous date on Γ and on those portions of the distance AB where a(x,0) < 1 applies. There are 2 Slavic references.

Card 2/3

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An Elliptical Type of Equation which Degenerates on the Boundary of the Do-

ASSOCIATION:

Tiflis Mathematical Institute AN Georgian SSR im.A.M.Ramadze (Tbilisskiy matematicheskiy institut imeni A.M.Ramadze Aka-

demii nauk GruzSSR) PRESENTED:

March 8, 1957, by M.A. Lavrentiyev, Academician

SUBMITTED: March 4, 1957

AVAILABLE: Library of Congress

Card 3/3

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16(1)-16.3500

SOV/20-129-2-11,66

AUTHOR:

Tersenov, S.A.

On a Hyperbolic Equation Degenerating on the Boundary

PERIODICAL: Doklady Akademii nauk SSSR,1959,Vol 129,Nr 2,pp 276-279(USSR)

ABSTRACT:

Let

(1) $L(u) = y^n u_{yy}^{-u} + au_y + bu_x + cu = f$ be given, where a,b,c,f are analytic in x and $y \ge 0$.

 $x \pm \frac{2}{2-n} y^{1-n/2} = c$ (2)

are the characteristics of (1). Let the domain D of the upper half plane be bounded by two characteristics, by the point (ξ_0, η_0) and by the interval AB of the x-axis, where A and B

are the points of intersection of the characteristics with Theorem 1. α .) If a(x,0)<1 on AB, then there exists a the x-axis. Let n=1. single solution u(x,y) of (1) in D satisfying the conditions

 $u(x,0) = \mathcal{T}(x), \quad \lim_{y\to 0} \frac{1}{\eta_y} \left[u_y^{-w} y \right] = \omega(x).$

Card 1/2

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